# 5/H-23 (vi) (Syllabus-2019)

### 2022

(November)

**CHEMISTRY** 

( Honours )

(Chem-H-502)

(Organic Chemistry—V)

Marks: 37

Time: 2 hours

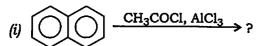
The figures in the margin indicate full marks for the questions

- 1. (a) Explain why electrophilic substitution reaction in naphthalene occurs preferentially at C1.
  - (b) Give a suitable method for the preparation of 1-naphthol.

(Turn Over)

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(c) Complete the following reactions:  $1 \times 3 = 3$ 



(iii) 
$$+ HNO_3 \frac{(CH_3CO)_2O}{\Delta}$$

- (d) Draw the structure of vitamin  $A_1$ . 2
- (e) Write the structure of the tripeptide Gly-Ala-Phe.

OR

2. (a) Complete the following reaction with mechanism:

(b) Write the products of the following reactions: 1×2=2

(i) 
$$\underbrace{\text{Na}_{2}\text{Cr}_{2}\text{O}_{7}}_{\text{H}_{2}\text{SO}_{4}} \rightarrow ?$$

(ii) 
$$\longrightarrow$$
 PNO<sub>2</sub> ?

- (c) Write a note on the secondary structure of proteins. 1½
- (d) Give a suitable method for the synthesis of vitamin C. 2½
- (e) Draw the structure of β-carotene. 1

3. (a) Assign the R- and S-configuration of the following:  $1 \times 2 = 2$ 

- (b) Draw the chair conformers of *trans*-1,2-dimethylcyclohexane. Explain which one is more stable and why.
- (c) Give one example each for cumulated and isolated diene. 1+1=2
- (d) Write short notes on the following: 1+1=2
  - (i) Urea-formaldehyde resin
  - (ii) Polyurethane

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(Continued)

(Turn Over)

3

#### OR

- **4.** (a) Define relative and absolute configurations.
  - (b) State the key differences between addition polymerization and condensation polymers.
  - (c) Convert the Fischer projection formula of the following molecule to Newman projection formula:

- (d) Discuss the molecular orbital structure of 1,3-butadiene.
- (e) Write short notes on the following:
  - (i) Neoprene
  - (ii) Dacron
- 5. (a) What are ketals? Write the mechanism for the acid-catalyzed reaction of cyclohexanone with ethylene glycol. 1+2=3

(b) Predict the product of the following reactions with mechanism: 2+2=4

(i) 
$$C-H + CH_3-C-H \xrightarrow{aq.NaOH}$$
?

ii) 
$$\langle \bigcirc \rangle$$
 + CO + HCI  $\xrightarrow{\text{AlCl}_3} ?$ 

(c) Write the products of the following reactions:  $1\frac{1}{2}\times2=3$ 

(i) 
$$\langle \bigcirc \rangle$$
-NH-NH- $\langle \bigcirc \rangle$ - $\overset{H^+}{\longrightarrow}$ ?

## OR

**6.** (a) How do you synthesize the following two compounds? Give the mechanism:

2×2=4

(i) 
$$C_6H_5$$
 C=C COOH COOH

2

2

1

2

(b) Write the products with mechanism: 2×2=4

(ii) 
$$Ph_3P=CH_2 \longrightarrow ?$$
(ii)  $NH_2OH \longrightarrow ?$ 

(c) Write the mechanism and the reagent for the following reaction: 1+1=2

- 7. (a) Write the synthesis of quinoline using Skraup synthesis with mechanism.
  - (b) What is the most likely position of substitution by Br<sub>2</sub> on the indole ring? Explain with mechanism.
  - (c) What are solid-state reactions? Give example.

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(Continued)

- (d) Write the preparation of any two of the following: 1×2=2
  - (i) Lead tetraacetate
  - (ii) LiAlH4
  - (iii) KMnO<sub>4</sub>

### OR

- (a) Write any three basic principles of Green chemistry.
  - (b) Write the products of the following reactions: 1×3=3

3

1

(ii) 
$$\bigcirc N \xrightarrow{C_6H_5MgBr} ?$$

- (c) Write short notes on the following:  $1 \times 2 = 2$ 
  - (i) Microwave-assisted reactions
  - (ii) K2Cr2O7
- (d) What do you mean by atom economy reaction? Give example.

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