

4/EH-24 (iv) (Syllabus-2020)

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(May/June)

PHYSICS

(Elective)

**(Special Theory of Relativity, Quantum
Mechanics—I, Atomic Physics—I, Nuclear
Physics—I and Solid State Physics—I)**

[PHY-04 (T)]

Marks : 56

Time : 3 hours

*The figures in the margin indicate full marks
for the questions*

Answer any eight questions

1. Describe the Michelson-Morley experiment and discuss the various interpretations given to the negative result obtained therefrom. 7
2. (a) Deduce Einstein mass-energy relation $E = mc^2$ and give its physical significance. 3+1=4

(2)

- (b) Derive the formula $E^2 = m_0^2 c^4 + p^2 c^2$ where the symbols have their usual meanings. Hence obtain relativistic relation between kinetic energy and momentum of a particle. 2+1=3
3. (a) Mention any two of the failures of classical physics which led to the development of quantum theory. 2
- (b) State Heisenberg uncertainty principle. Illustrate this principle by a single-slit electron diffraction experiment. 1+4=5
4. Solve the Schrödinger equation for a particle enclosed in a one-dimensional box of side L and obtain its energy values. Draw a graph of its first three wave functions. Discuss the probability of finding the particle at different points in the box when it is in different states. 4+1+2=7
5. Describe Thomson's method for the determination of e/m of an electron. Why is the measurement of this ratio important? 6+1=7
6. (a) Discuss, with a neat diagram, the construction of Bainbridge mass spectrograph. 3

(3)

- (b) Name the quantum numbers that help to specify the energy state of the electron in an atom. Give their significance. 2+2=4
7. (a) Briefly describe how X-rays are produced in the Coolidge tube. 3
- (b) What are characteristic X-rays? Distinguish between continuous X-ray and characteristic X-ray emission spectra. 1+3=4
8. What is natural radioactivity? Give a brief account of the nature and properties of the radiations emitted in a radioactive decay. 1+6=7
9. What are cosmic rays? Discuss the altitude effect and latitude effect of cosmic rays. 1+6=7
10. (a) What do you mean by a chain reaction in nuclear fission? Explain controlled and uncontrolled chain reactions. 1+2=3
- (b) Show that there is a critical size of the fissionable material in order that the chain reaction may be sustained. 4

11. Define coordination number. Compute the same for simple cubic, b.c.c. and f.c.c. lattices. Also prove that the distances between the nearest neighbours are $a, \frac{a\sqrt{3}}{2}$ and $\frac{a}{\sqrt{2}}$ respectively. 1+3+3=7

12. (a) What is superconductivity? Discuss the experimental evidence of the occurrence of superconductivity in metals and alloys. 1+3=4

- (b) Distinguish between type-I and type-II superconductors. 3
