2/H-77 (ii) (Syllabus-2015)

2016

(April)

BIOTECHNOLOGY (Honours)

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(Biological Chemistry)

Marks : 56

Time: 3 hours

The figures in the margin indicate full marks for the questions

Answer Question No. 1, which is compulsory and four from the rest

	and lour nom are real						
1.	(a)	What do you understand by electrochemical gradient?	2				
	(b)	What general properties do fatty acids	2				
	(c)	Give the steps involved in glycogen	2				
	(d)	What are the various ways by which	2				
	(e)	Is gluconeogenesis a reversal of	2				
	(f)	What are symport and antiport?	2				

2.	(a)	addition of small amounts of acid or		6.	(a)	How is palmitate oxidised to form acetyl CoA?		
		base." Explain how buffers resist this change in pH when (i) an acid and (ii) a			(b)	Discuss the essential differences between the C3, C4 and the CAM		
	(h)	base are added to it.	5			pathways.		
	(b)	Derive Henderson-Hasselbalch equation.	6			otrono Albain		
3.	(a)	What is entropy?	3	7.	On	n what basis are enzymes given their ames? Explain in detail the nomenclatural		
	(b)	"Entropy of the universe is always increasing." Comment.	3		pro	cedure giving suitable examples.		
	(c)		5	8.	(a)	Write a brief account on enzyme isolation and purification. What are the criteria of purity?		
4.	(a)							
	• •	Give the structure of sucrose. Why is it called a non-reducing sugar? 2+	2=4		(b)	Explain the role of zymogens with		
	(b)	Why are α-amino acide as collod?				one example.		
		platif with a diagram.	2			***		
	(c)	what are the forces that stabilize quaternary structure of protein 2 Civa						
		quaternary structure.	5					
5.	(a)	Trace the route taken by an electron in the inner mitochondrial membrane starting from NADH and ending at oxygen. Illustrate using suitable diagrams.	7					
	(b)		7					
	1-7	Explain the regulation of glycolysis with respect to the irreversible steps of the pathway.						
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